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JOU&RNAL

OF THE

AMERICAN CHEMICAL SOCIETY

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CONTENTS

PHYSICAL CHEMISTRY

	. *	and the state of t	
Ralph K. Birdwhistell and Ernest Griswold: The Effect of Some Salts on the Solubility of Silver Acetate and of Silver Nitrate in Acetic Acid at 30°. R. K. Brinton: The Photolysis of Acetaldazine. James P. Coughlin: High-Temperature Heat Contents, Heat of Transition, and Heat of Fusion of Anhydrous Sodium Sulfate. Isaac Feldman, T. Y. Toribara, Jean R. Havill and W. F. Neuman: The Beryllium-Citrate System. II. Ion-etchange Studies. Paul Yen-hsiung Feng and Joseph W. Kennedy: Electrical and Chemical Effects of β-Radiation	873 842 868 878	 R. Lee Purlee, Robert W. Taft, Jr., and C. A. De-Fazio: Enthalpies and Entropies of Activation for the Hydration of Dissolved Isobutene and Trimethylethylene from the Thermodynamic Properties for Solution of Gascous Olefins in Aqueous Nitric Acid. F. F. Rawlings and E. C. Lingafelter: X-Ray Crystallography of the Sodium n-Alkyl Sulfates Notes Fred H. Coats and Robbin C. Anderson: Electron Impact Data on Substituted Acetylenes: Propyne and 2-Butyne. 	837 870 895
in Polystyrene	847	Walter Dannhauser and Philip A. Vaughan: The	
Harold Hart and William L. Spliethoff: Kinetics of	**	Crystal Structure of Cuprous Chromite	896
the Racemization of a Phenethyl Chloride in	۸.	O. J. Kleppa: The Heat of Formation of InSb	
Phenois		H D Owen and O A Sabandara The Tanking at	897
Lucious, and a second s	833	H. R. Owen and O. A. Schaeffer: The Isotope Abun-	- S
Norman C. Li, Ting Li Chu, Charles T. Fujii and		dances of Chlorine from Various Sources	898
Janes M. White: Association of Imidazole with	ার ্ব	Ralph P. Seward: The Conductance and Viscosity	; s.
Nickel(II) and Alkaline Earth lons.	859	of Highly Concentrated Aqueous Solutions of	
William F. Linke: The System Magnesium Bromate-	.005.	Hydrazinium Chloride and Hydrazinium Ni-	
Motor	'	trate	905
Water	866	Robert W. Taft, Jr., E. Lee Purlee and Peter Riesz:	9 000
R. S. Mulliken: Structures of the Halogen Molecules	4 .,	A Method for Determining the Distribution	1
and the Strength of Single Honds	884	Constant for a Columnia to the Distribution	
R. S. Mulliken: Bond Angles in Water-Type and	y .	Constant for a Substance between the Gas Phase	Ł.
Ammonia Type Molecules and their Derivatives.	887	and a Condensed Phase	899
Robert T. O'Connor, Robert R. Mod, Mildred D.	007	Robert W. Taft, Jr., and Peter Riesz: Thermo-	
Miren and Prold I Chair of the D		dynamic Properties for the System Isobutene-	
Murray and Evald L. Skau: The X-Ray Diffrac-		-Butyl Alcohol	902
tion and Infrared Spectra of Molecular Com-		J. R. Tomlinson, L. Domash, R. G. Hay and C. W.	
pounds of Acetamide and Long-Chain Saturated		Montgomery: The High Temperature Heat	
Fatty Acids.	892	Content of Nickel Oxide	900
			74,0
INOPGA	NIC (CHEMISTRY	
	711C F	Memisiki.	
Gilbert H. Ayres and Max H. Booth: Catalytic De-	* 5	Volumes and Electrical Conductivities of the	
composition of Hypochlorite Solution by Iridium		Moltan Custom, Matabase District of the	
	825	Molten System: Molybdenum Trioxide Sodium	
Gilbert H. Ayres and Max H. Booth: Catalytic	.7	Molybdate	851
Decomposition of Hypochlorite Solution by		Alexander I. Popov and Wesley W. Wendlandt: The	
	4	Methylamine Complexes of the Rare Earth (III)	
William W. Trans. Common W. Col. R. Rinetic Studies	828	Chlorides	857
William V. Hough, George W. Schaeffer, Marcelline		Walter C. Schumb and Walter J. Bernard: The Thio-	
Dzurus and Albert C. Stewart: . The Preparation,	3	chlorides of Silicon.	000
Identification and Characterization of the N-		A TO Constitution of the same	862
Trialkylborazoles	864	A. K. Sundaram and B. B. Sandell: Chloro Com-	4 . 4
E. H. Huffman, G. M. Iddings, R. N. Osborne and	7.4	plexes of Palladium(II) in Solution	3855 °
G. V. Shalimoff: Extraction of Zirconium and			
	881 ^	Notes	
M. H. Lietzke and J. V. Vaughen: The Behavior of	001		•
the Silver, Silver Chiloride and the Mercury,		Dietmar Seyferth and Eugene G. Rochow: The	
Mamurana Chlorida Planta de 177 1	į.,	Preparation of Chloromethyl Derivatives of	. 1
Mercurous Chloride Electrodes at High Tem-		Germanium and Silicon by the Diazomethane	
peratures.	876	Method.	907
Kelso B. Morris, Marlene L. Cook, Clarice Z. Sykes		Walter C. Schumb, and Walter J., Bernard: The	
and Malcolm B. Templeman: Densities, Molal		Formation of Silicon Monosulfide.	904
· · · · · · · · · · · · · · · · · · ·			20.1

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 $(1-2 \times 10^8 \text{ MSH u./g.})$ was obtained. Eight grams was mixed with 150 ml. of 0.1 N acetic acid and centrifuged. Sixteen grams oxycellulose was added to the supernate and the mixture shaken for 75 min. Oxycellulose was removed by centrifugation, washed with 0.1 N acetic acid, then shaken in 100 ml. of 80% acetic acid for 60 min. The supernate was diluted with equal quantities of water. Lyophilization yielded 0.35–0.5 g. product (1–2 \times 10 9 MSH u./g.). One and a half grams of this fraction was distributed through a 12-tube countercurrent system at 5° using sec-BuOH and 0.5% aqueous trichloroacetic acid. The contents of tubes 4-6 were combined and lyophilized. Approximately 0.5 g. of solids $(3-4 \times 10^9 \text{ MSH u./g.})$ was obtained. Forty mg. was subjected to paper electrophoresis at 5°, 18 volts/cm., 8-10 hours, pH 8.9 using barbiturate-acetate-hydrochloric acid buffer (u = 0.056). Four components were visualized with 1% brom phenol blue staining. That moving fastest toward the cathode was extracted with 20% acetic acid and lyophilized. The product was dissolved in 1 ml. of 0.2 N acetic acid and subjected to paper electrophoresis at pH 4.9 using pyridine-acetic acid buffer (u=0.1), 5°, 18 volts/cm., 10-12 hours. Staining revealed a single component moving toward the cathode. The active area was extracted with 20% acetic acid and lyophilized. The white solid, 2.5 mg., $(1.5-2.5 \times 10^{10} \text{ MSH u./g.})$ represented about 30% of the total MSH activity placed on the first electrophoretic run at alkaline pH. Ninhydrin reaction of hydrolyzed extracts of different parts of the filter paper run at

TABLE				
Amino acid*	Per cent.13	Molecular ratious		
Aspartie	3.8	3		
Glutamic	5.3	3		
Serine	3.4	3		
Glycine	2.3	3		
Tyrosine	4.1	2		
Lysine	5.3	3		
Arginine	4.0	2		
Valine	2.9	, <u>2</u> 2		
Phenylalanine	4.7	3		
Alanine	1.4	1		
Cystine ^{to}	5.3	2		
Proline	3.7	.3		
Leucine	1.7	i		
Threonine	1.3	i		
Histidine	0.6	0		
Tryptophan ¹¹	• • •	2		
Total	49.8	34		

(9) Semi-quantitative amino acid analyses were done by A. M. Gross and W. F. White of the Research Department, Armour Laboratories using filter paper chromatography and determining the intensity of ninhydria stained areas with a densitometer: J. F. Rowland and A. M. Gross, Anal. Chem., 26, 502 (1954).

(10) Cyntine and cysteine are not distinguished in the analysis. However, cysteine is probably absent because MSH is not oxidized and reduced readily as would be expected were this amino acid present. Methiculus was not tested for.

(11) Tryptophan was determined by ultraviolet absorption after subtracting tyrosine from the total value; A. B. Lerner and C. P. Barnum, Arch. Biochem., 10, 417 (1946).

(12) Tryptophan, methionine, moisture and ash were not included in the total amino acid per cent, analysis.

(13) Molecular ratios are given in whole numbers and represent only approximate values. pH 4.9 showed MSH activity associated with the predominant color response.

The active fraction moved as a single component (staining with bromophenolblue) on paper electrophoresis at pH 1.4, 4.9, 8.9, 11.3 and 12.2. Since at pH 11.3 movement towards the anode was slight compared with dextran, the iso-electric pH was estimated to be in the region of 10.5-11. On the basis of amino acid composition minimum molecular weight was estimated at 4500. MSH activity of the final product was approximately 500 times that of the original hog posterior pituitary powder with little ACTH activity. This fraction behaved as a single component when distributed in a 97 tube countercurrent apparatus employing the solvents described previously. Although the MSH preparation, assumed to be a polypeptide, was tested by electrophoresis and countercurrent distribution, other criteria for homogeneity remain to be satisfied.

(14) MSH has little if any vasopressin or ACTH activity. Assays set to detect one unit each of ACTH or vasopressin, using 378 and 449 mcg. of MSH, respectively showed no activity.

DIVISION OF DERMATOLOGY AARON B. LERNBR UNIVERSITY OF OREGON MEDICAL SCHOOL TEH H. LEE PORTLAND 1, OREGON

RECEIVED JANUARY 13, 1955

A NEW METHOD OF FORMING PEPTIDE BONDS Sir.

We wish to describe a new and very useful method of forming peptide or other amide bonds. The two components, one containing a free carboxyl function and the other a free amino group, couple directly and rapidly in high yield on treatment with N,N'-dicyclohexylcarbodiimide at room temperature.

In contrast to other schemes for carboxyl activation involving mixed anhydride formation, the reaction is not sensitive to moisture; indeed, it may be carried out in aqueous solution. The remarkable selectivity of the reagent is attested by the successful use of carbobenzoxyserine as an acylating moiety without protection of the hydroxyl group. No racemization was detected employing as the acylating agent a dipeptide derivative in which an optically active amino acid furnished the free carboxyl function (carbobenzoxyglycyl-r-phenylalanine), an observation of considerable importance in the synthesis of larger peptides by joining units containing two, three or more amino acids. The co-product, N.N'-dicyclohexylurea, has a very low solubility in most organic or aqueous solvents, and, in all cases tried, is easily separated.

RCO₂H + NH₂R' + C₄H₁₁N=C=NC₄H₁₁ \longrightarrow RCONHR' + C₄H₁₁NHCONHC₄H₁₁

The simplicity, convenience and efficiency of this technique may be illustrated by the synthesis of a tripeptide derivative. After a 4-hour period at room temperature, a solution in tetrahydrofuran of carbobenzoxyglycyl-L-phenylalanine containing a slight excess of crystalline N,N'-dicyclohexyl-carbodiimide and ethyl glycinate was treated with a small amount of acetic acid (to decompose the

(1) Readily prepared by the method of R. Herbeck and M. Pezzuti, Ber., 71, 1933 (1938).

excess reagent). The insoluble urea was removed, the solvent was replaced by ethyl acetate, and the solution was washed with dilute acid and aqueous potassium bicarbonate. The addition of petroleum ether afforded 87% of crystalline carbobenzoxyglycyl-L-phenylalanylglycine ethyl ester; m.p. 118-119°, $[\alpha]^{27}p - 13.5$ ° [ethanol] (reported²: m.p. 116-118°, $[\alpha]^{25}p - 12$ °). In a similar fashion we have prepared a variety of dipeptide derivatives,

including the following examples.

In methylene chloride, phthaloyl-L-phenylalanylglycine ethyl ester was produced in 92% yield; m.p. 161–162°, $[\alpha]^{25.6}$ p -146°, (reported³: m.p. 161–162°, $[\alpha]^{25.5}$ p -146°). In aqueous tetrahydrofuran, a product of the same quality was obtained in 72% yield. Phthaloyl-L-alanyl-L-proline benzyl ester (74%) was isolated with m.p. 101-102°. $[\alpha]^{28.5}$ D -135° [ethanol]. Anal. Calcd. for $C_{23}H_{22}N_2O_6$: C, 67.98; H, 5.42; N, 6.90. Found: C, 68.07; H, 5.52; N, 6.77. Carbobenzoxy-L-serine and ethyl glycinate coupled to give carbobenzoxy-L-serylglycine ethyl ester (59%) in tetrahydrofuran: m.p. $106-107^{\circ}$, [ethanol], reported, m.p. $105-107^{\circ}$. Phthaloyl-L-phenylalanyl-L-leucine ethyl ester (91% yield) had a m.p. of $109-110^{\circ}$, [α]^{25, 4}p. -115° [ethanol]. Anal. Calcd. for C-H-NO-C 68 78: H 6.47: N 6.42. Found for C₂₅H₂₈N₂O₅: C, 68.78; H, 6.47; N, 6.42. Found: C, 68.50; H, 6.59; N, 6.48.

DEPARTMENT OF CHEMISTRY JOHN C. SHEBHAN MASSACHUSETTS INSTITUTE OF TECHNOLOGY CAMBRIDGE 39, MASSACHUSETTS GEORGE P. HESS RECEIVED JANUARY 11, 1955

9a-HALO-118-HYDROXY AND 11-KETO DERIVATIVES OF PROGESTERONE, DESOXYCORTICOSTERONE AND 17a-HYDROXYPROGESTERONE

Sir:

In previous communications^{1,2} there have been described the synthesis of 9α -halogenated derivatives of cortisone and hydrocortisone and shown that the glucocorticoid activity of these substances increased with decreasing atomic weight of the halogen atom. The most active member of that series, 9\alpha-fluorohydrocortisone acetate possessed about 11 times the activity of cortisone acetate in the rat liver glycogen assay. Soon thereafter it was found that in addition to being potent glucocorticoids these compounds were highly effective in controlling electrolyte balance and in maintaining life in the rat*, dog45 and in man. 4.6

It appeared of great interest to ascertain what influence variations in the side-chain might have upon the adrenocorticoid activity of such halogen-

ated derivatives. For this purpose we have prepared the 9α -halo derivatives (halogen = Br, Cl, F) of 11β -hydroxyprogesterone, 11β , 17α -dihydroxyprogesterone and corticosterone acetate and of the corresponding 11-ketones by a synthetic route paralleling that described in our earlier publications. 1.2 This synthesis proceeds from the 11mesylates of the requisite 11α-hydroxy derivatives⁷ (11α-hydroxyprogesterone mesylate, m.p. 165-167°; $[\alpha]^{28}D + 135°$ (c, 0.77 in CHCl₂); $\lambda_{\text{max}}^{\text{alc}}$ 238 $m\mu$ ($\epsilon = 17,200$); Anal. C, 64.81; H, 7.63; S, 7.48. Epicorticosterone 11α-mesylate 21-acetate, m.p. $156-157^{\circ}$; $[\alpha]^{23}D + 144^{\circ}$ (c, 0.92 in CHCl_s); λ_{\max}^{alc} 238 m μ ($\epsilon = 16,600$); Anal. C, 61.52; H, 7.07. 11α , 17α -Dihydroxyprogesterone 11α -mesylate, m.p. 150-152°; [a]23D +64° (c, 0.49 in CHCl₃); $\lambda_{\text{max}}^{\text{alc}}$ 238 m μ ($\epsilon = 18,200$); Anal. C, 62.11; H, 7.71; S, 7.11), via the 9,11-unsaturated steroids (9(11)-dehydro- 17α -hydroxyprogesterone, 214–216°; $[\alpha]^{23}D + 67^{\circ}$ (c, 0.82 in CHCl₃); $\lambda_{\text{max}}^{\text{alc}}$ 239 m μ (ϵ = 18,450); Anal. C, 76.52; H, 8.46), to the 9α ,11 β -bromohydrins (see table). The latter on treatment with base yielded the 98,118-epoxides $(9\beta,11\beta$ -oxidoprogesterone, amorphous, $[\alpha]^{23}D+61^{\circ}$ (c, 1.55 in CHCl₃); $\lambda_{\text{max}}^{\text{alc}}$ 243 m μ (ϵ 13,600). 9 β ,11 β -Oxidodesoxycorticosterone acetate, m.p. 137-138°; [α]²²D +61° (c, 0.66 in CHCl₃); $\lambda_{\text{max}}^{\text{alc}}$ 243 m μ (ϵ = 15,100); Anal. C, 71.81; H, 8.10. 98,11 β -Oxido-17 α -hydroxyprogesterone, m.p. 183–184°; $[\alpha]^{22}$ D -32° (c, 1.02 in CHCl₃); $\lambda_{\text{max}}^{\text{alc}}$ 243 m μ ($\epsilon = 16,600$); Anal. C, 72.99; H, 8.11), which upon reaction with the requisite hydrogen halides formed the 9α -chloro- and 9α -fluoro-11 β -hydroxy derivatives. Oxidation with chromic acid furnished the corresponding 11-ketones. Alternatively, the 9α-chloroderivatives could be prepared by allowing the 9(11)unsaturated steroids to react with N,N'-dichlorodimethylhydantoin in the presence of perchloric

The physical properties of the halogenated steroids and the activities of representative compounds in the liver glycogen and sodium retention assays in the adrenalectomized rat are listed in the accompanying table. As had been observed previously in the 9α -halohydrocortisone series both gluco- and mineralocorticoid activities were found to increase with decreasing atomic weight of the halogen atom. No significant differences were noted between the activities of the 11\beta-hydroxy and 11-keto derivatives. Outstanding among the compounds tested were 9α-fluoro-11β-hydroxy and 11-ketoprogesterone, which although lacking both the 17- and 21-hydroxyl groups approximately equalled cortisone acetate in glucocorticoid activity. The most potent mineralocorticoids of this series were 9\alpha-fluorocorticosterone acetate and 9\alpha-

(7) J. Fried, R. W. Thoma, J. R. Gerke, J. E. Herz, M. N. Donin and D. Perlman, Tens Journal, 74, 3962 (1952).

(8) The course of this reaction was dependent on the nature of the side chain. Thus, 9(11)-dehydro-17a-bydroxyprogesteroze assorted the desired chlorohydrin in about 50% yield. On the other hand, treatment of 9(11)-dehydroprogesterone with N,N'-dichlorohydantoin resulted in a mixture containing more than one atom equivalent of chlorine from which 9a-chloro-11\$-bydroxyprogesterone could be isolated only after reduction with chromous chloride. It appears likely that the extra chlorine atom reducible by chromous chloride is located in the 17-position.

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⁽⁵⁾ Aided by a fellowship from the National Poundation for Infantile Paralysis.

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(4) G. W. Liddle, M. M. Pechet and F. C. Bartter, Science, 120,

⁽⁵⁾ W. W. Swingle, C. Baker, M. Bisler, S. J. Le Brie and L. J. Brannick, Proc. Soc. Rxp. Biol. Med., In press.

⁽⁶⁾ A. Goldfien, G. W. Thorn, P. M. Beigleman and J. C. Luidlaw, J. Clin. Endocrinology, 14, 782 (1954).